Solution to the following complex reaction scheme with

$k_1 = 0.15\ min^{-1}$
$k_2 = 0.14\ min^{-1}$
$k_3 = 0.11\ min^{-1}$
$k_4 = 0.12\ min^{-1}$
$k_5 = 0.08\ min^{-1}$

$c_{A\circ} = 100$ mmol/L
$c_{B\circ}, c_{C\circ}, c_{D\circ}, c_{E\circ} = 0$ mmol/L